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1. ____ The nuclear equation is an example of an equation that

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represents (a) alpha emission. (b) beta emission. (c) positron emission. (d) electron capture. 2. _____ When undergoes electron capture to form a new element X, which of

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the conversion of atoms of one element to atoms of another. (It involves a nuclear change, not a chemical change)

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Nuclear Chemistry Review Answers Chapter Modern Chemistry 175 Nuclearchemistry CHAPTER 21 REVIEW Nuclear Chemistry SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. Match each of the following statements with the process (es) to which they apply, using one of the choices below: (1) fission only (3) both fission and ...

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answer choices. The amount of time it takes for some of the nuclei in a sample of the isotope to decay. The amount of time it takes for half the electrons in a sample of the isotope to decay. The amount of time it takes for half the nuclei in a sample of the isotope to decay.

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Nuclear Chemistry Nuclear Transformations • Rutherford in 1919 performed the first nuclear transformation. • The transmutations are sometimes represented by listing in order, the target nucleus, the bombarding particle, the ejecting particle and the product nucleus. • The above equation becomes: ${}_{11}^{24}\text{Na} + {}_2^4\text{He} \rightarrow {}_{13}^{27}\text{Al} + {}_0^1\text{n}$

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