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A Textbook of Chemical Engineering Thermodynamics. Author. K. V. Narayanan. Publisher. PHI Learning Pvt. Ltd., 2004. ISBN. 8120317327, 9788120317321. Length. 520 pages.

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, the solvation free energy of chemical species can then be obtained from any PVT EOS as: (15)  $\Delta G_i^{\text{sol}} = k T \int V = \infty V \int V \mathbf{1} - \partial N z \partial N i T, V d V$  We will illustrate this practice using the van der Waals equation of state later.

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